

Chapter 20

Spectrometers

Spectrometer Components

- source of radiation
- sample
- radiation selection device (monochromator)
- detector
- signal processor
- readout device

Components of Optical Instruments



Emission Flame Photometer

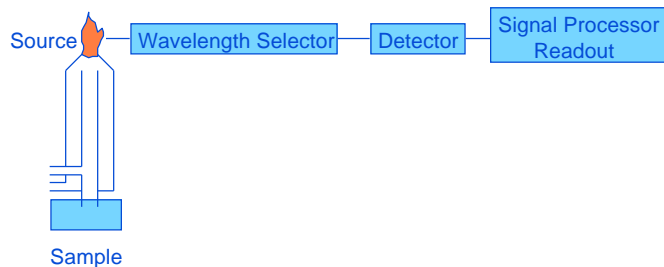
Flame Atomic Absorption Spectrometer

Absorption Spectrometer

Fluorescence and/or Scattering Spectrometer

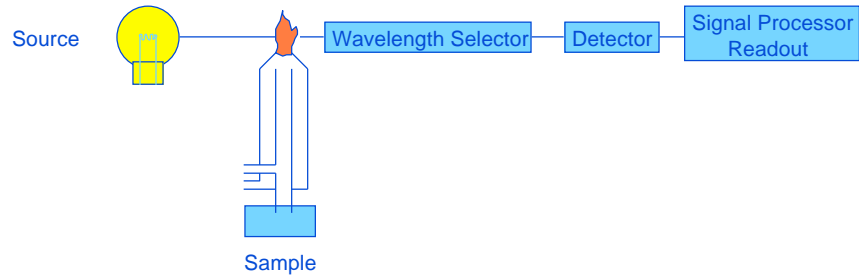
Components of Optical Instruments

Emission Flame Photometer



Components of Optical Instruments

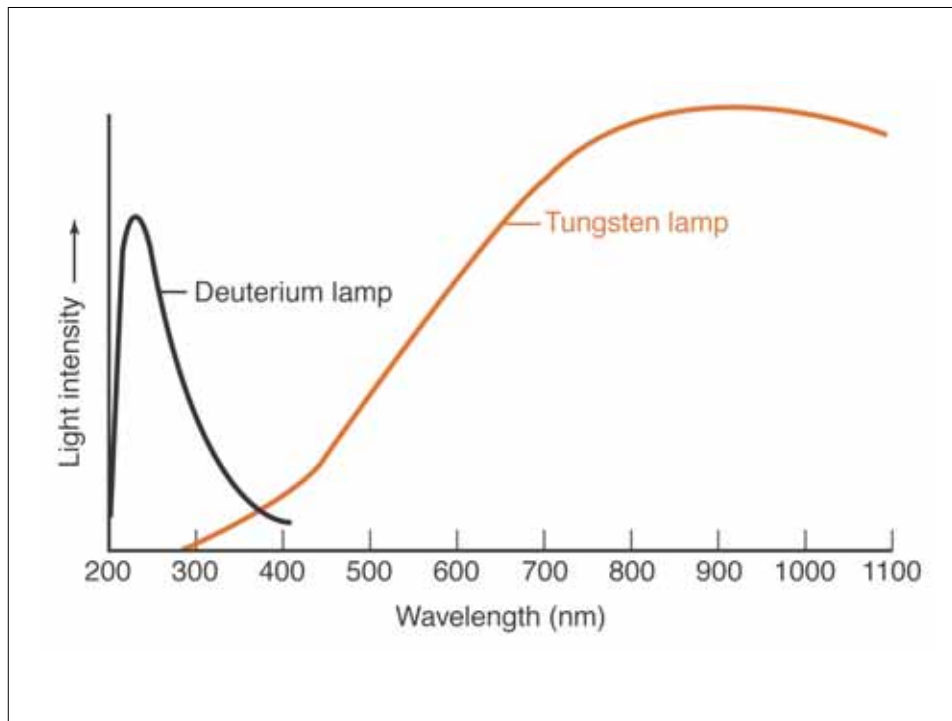
Flame Atomic Absorption Spectrometer



Components of Optical Instruments

Absorption Spectrometer





Sources

UV-Visible-Near IR Region

H₂ - D₂

160 - 375 nm, must use Quartz windows and cuvettes

Xe arc lamps

250 - 600 nm, max I at 500 nm

W filament

320 - 2500 nm, needs close V control

Sources

IR Region

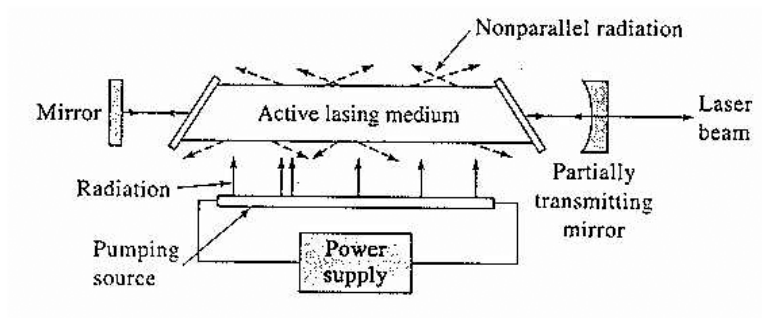
Nernst glower - rare earth oxides

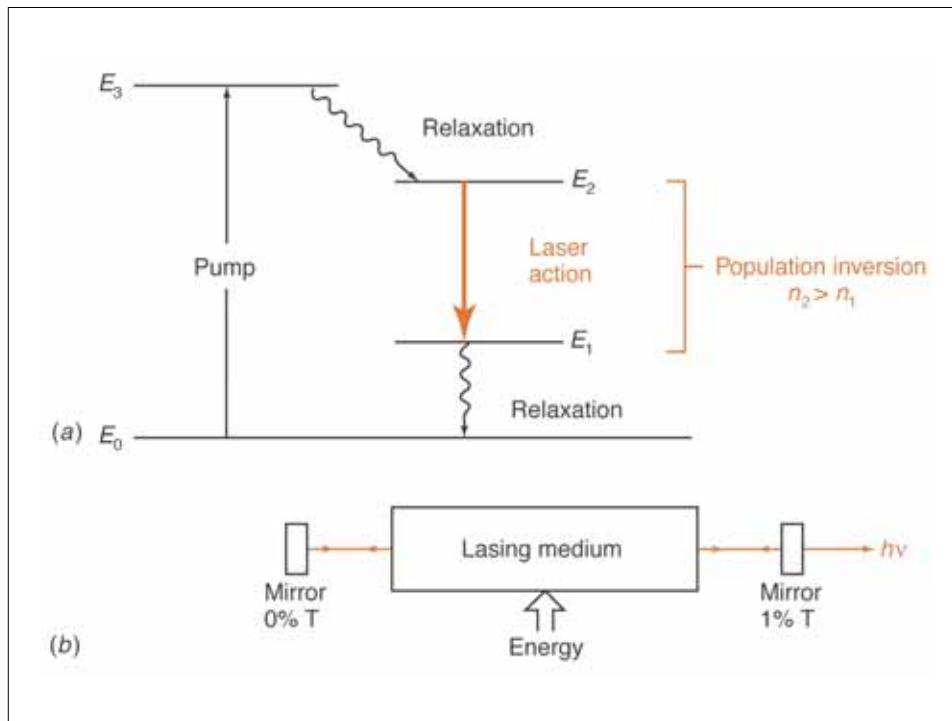
globalar - silicon carbide rod

incandescent wire - nichrome wire

Sources

Laser





Wavelength Selection

Filters

- interference filters
- interference wedges
- absorption filters

Wavelength Selection

Monochromators

Components

entrance slit

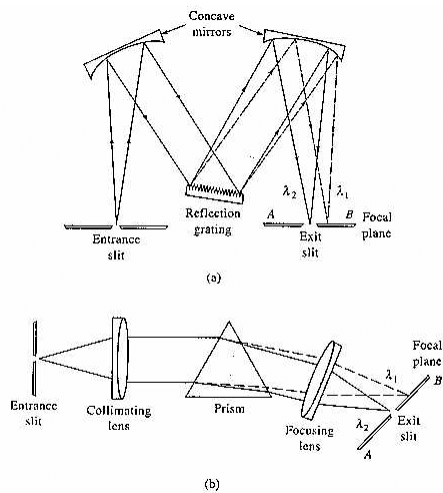
collimating element (lens or mirror)

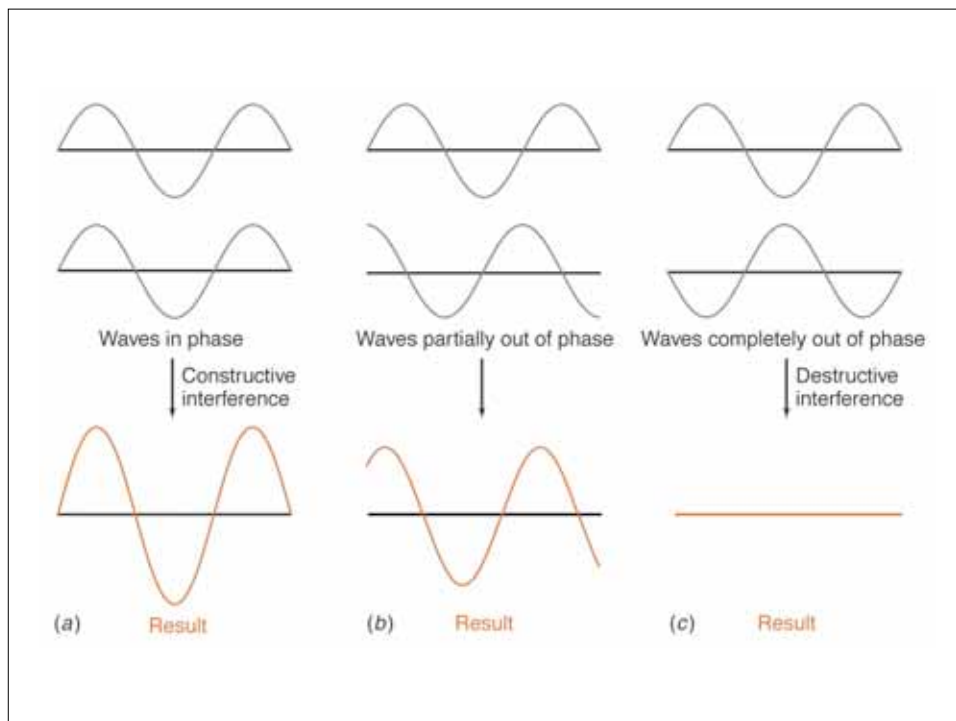
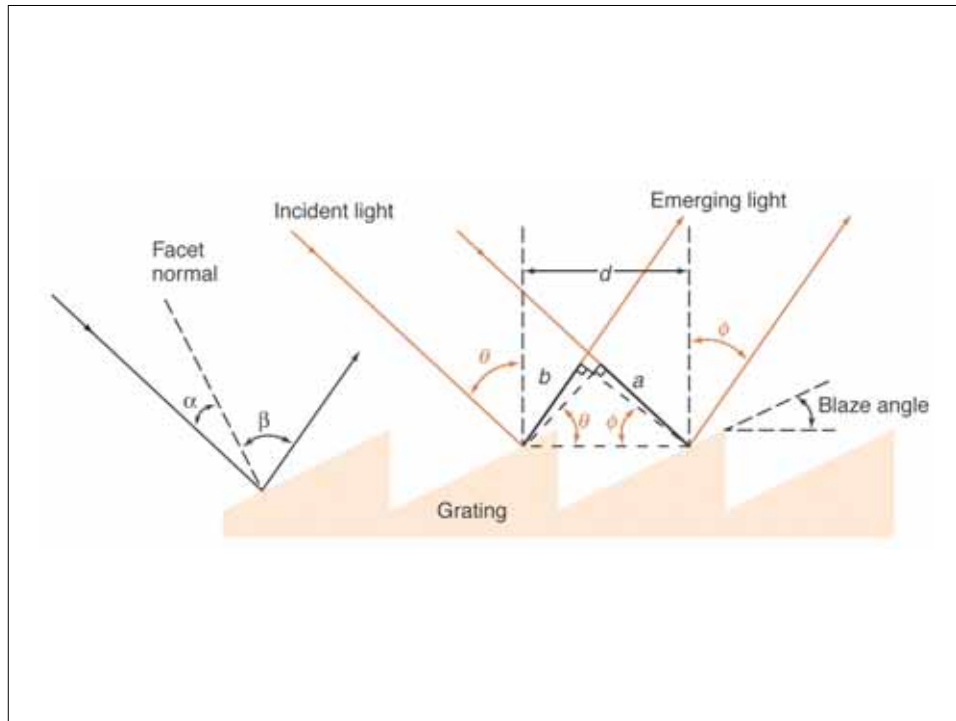
prism or grating as dispersing element

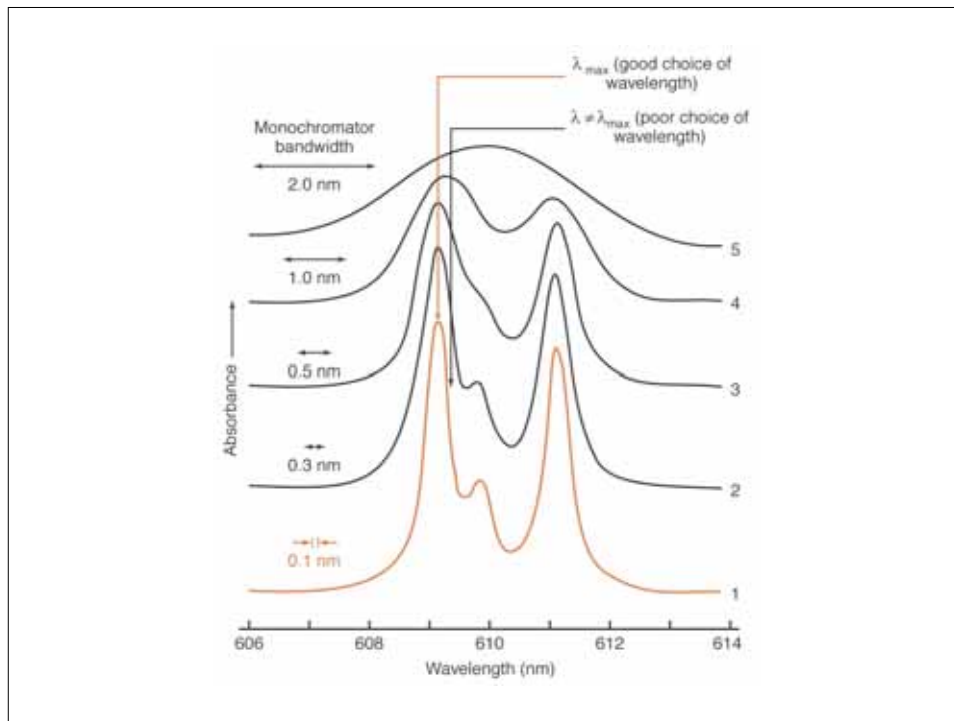
focusing element (lens or mirror)

exit slit

Grating vs. Prism







Sample Containers

Ultra-Violet

- quartz

Visible

- quartz
- glass

Infrared

- NaCl
- AgCl
- KBr

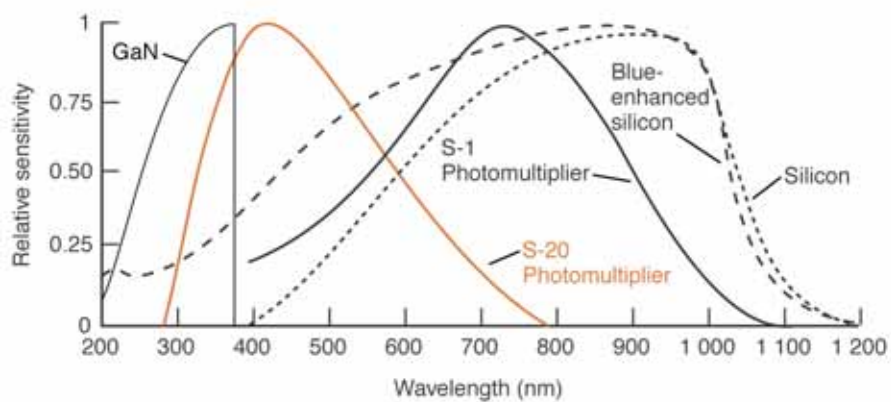
Types of Radiation Detection

UV-Visible

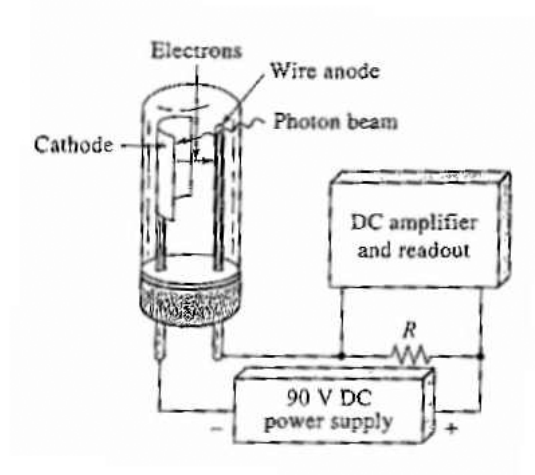
- Photon Detectors
- Vacuum Phototubes
- Photomultiplier Tubes
- Photodiodes
- Linear Photodiode arrays
- Charge-Transfer

Infrared

- Heat Detectors



Phototube



Photomultiplier Tube

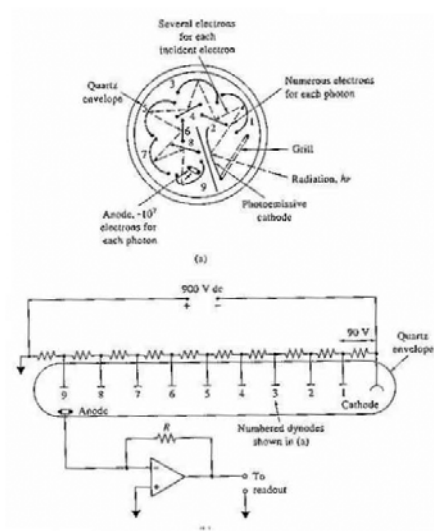
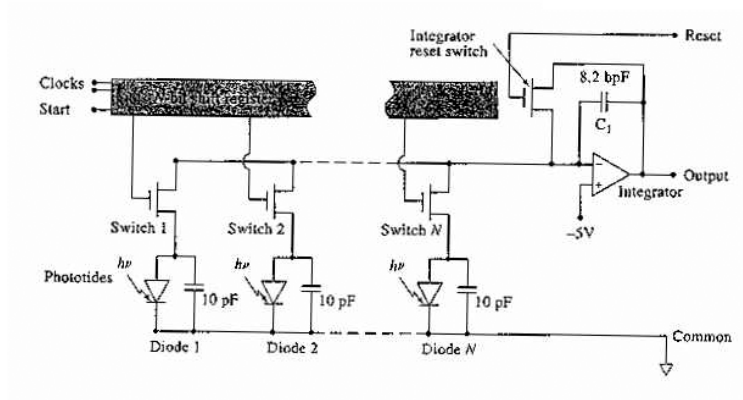


Photo Diode Array



Beer's Law Instrumental response

$$G = KP + K'$$



Beer's Law Instrumental response

$$G = KP + K'$$

dark-current adjust

$$G = KP + 0.00$$

$$K = G/P \implies G/P = 100/P_0$$



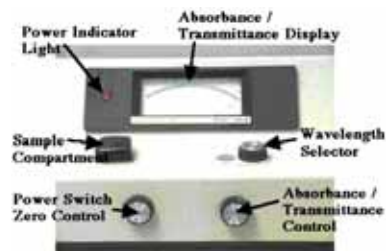
Beer's Law Instrumental response

$$G = KP + K'$$

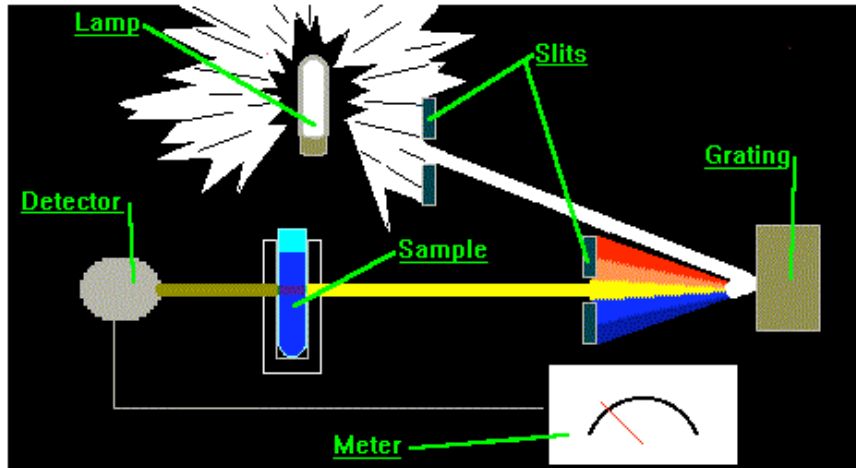
100% adjust, incident radiation

$$G_0 = 100 = KP_0 + 0.00$$

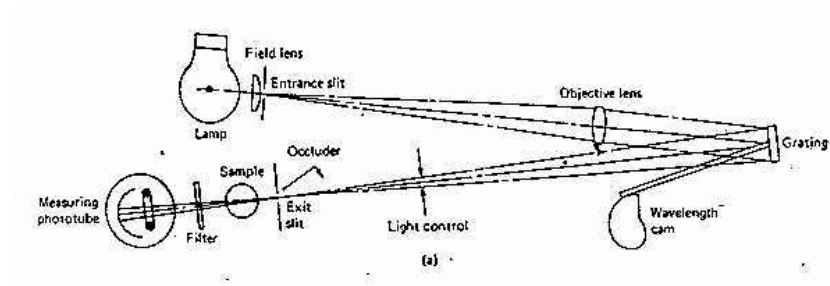
$$K = 100/P_0$$



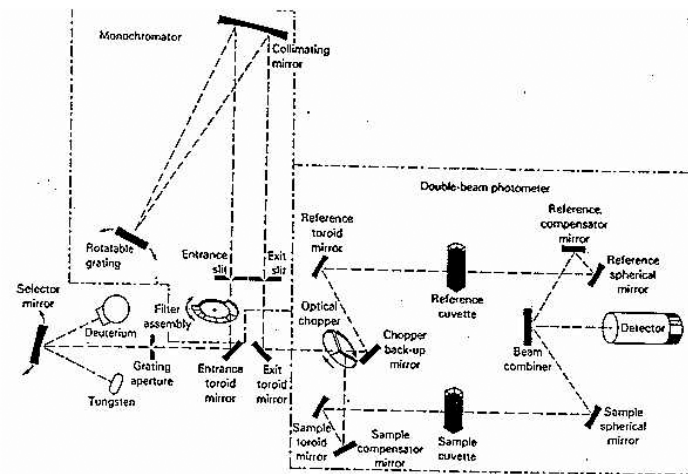
Spectronic 20 optical diagram



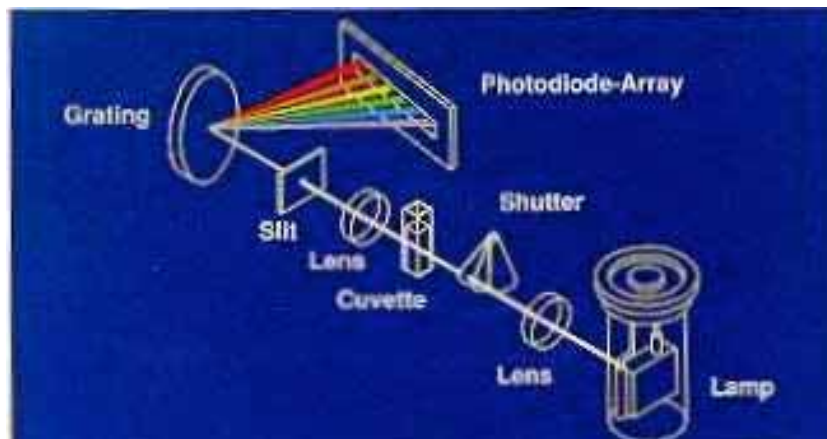
Spectronic 20



Double Beam Spectrometer



HP 8452a



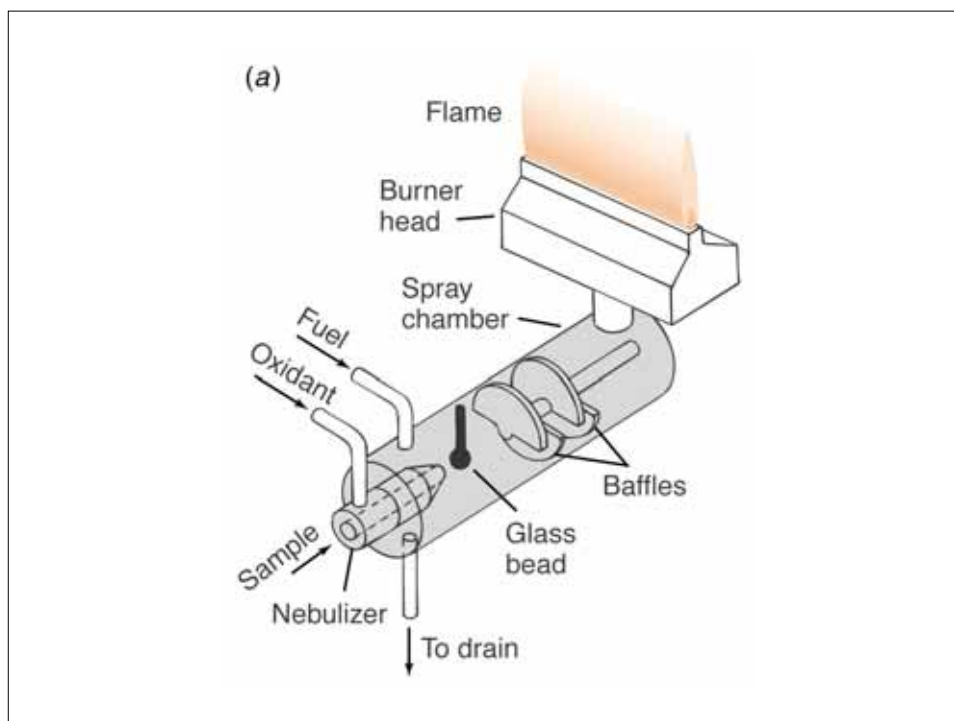
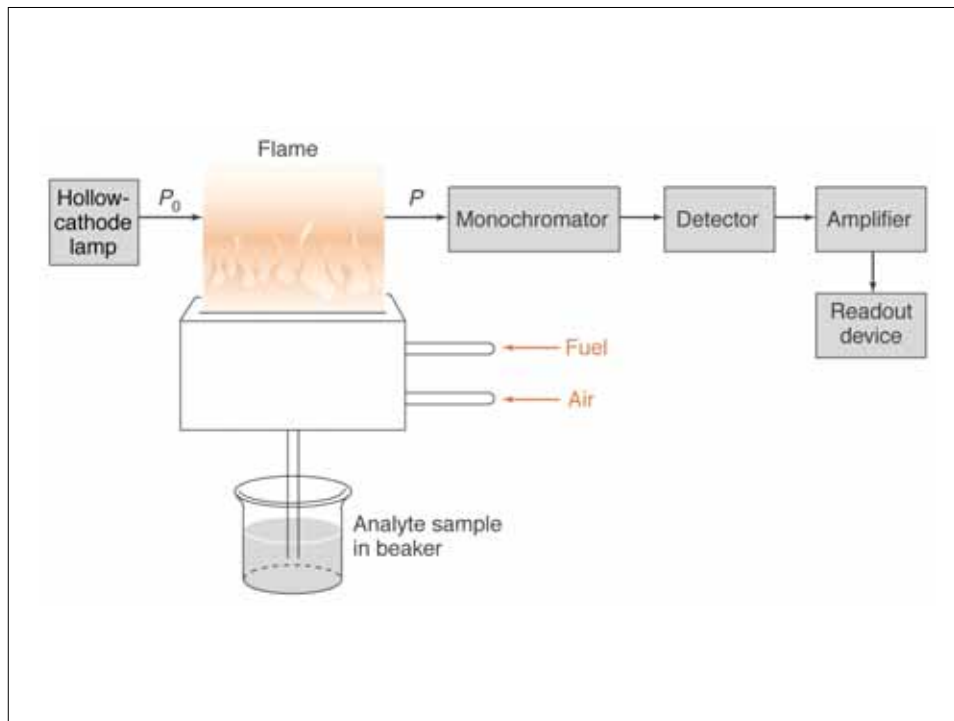
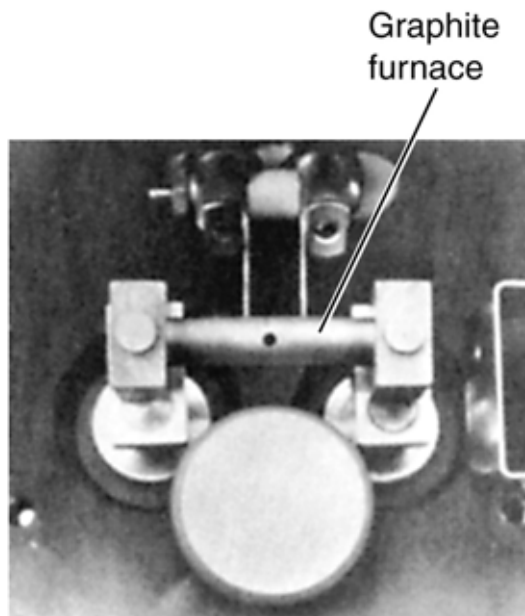
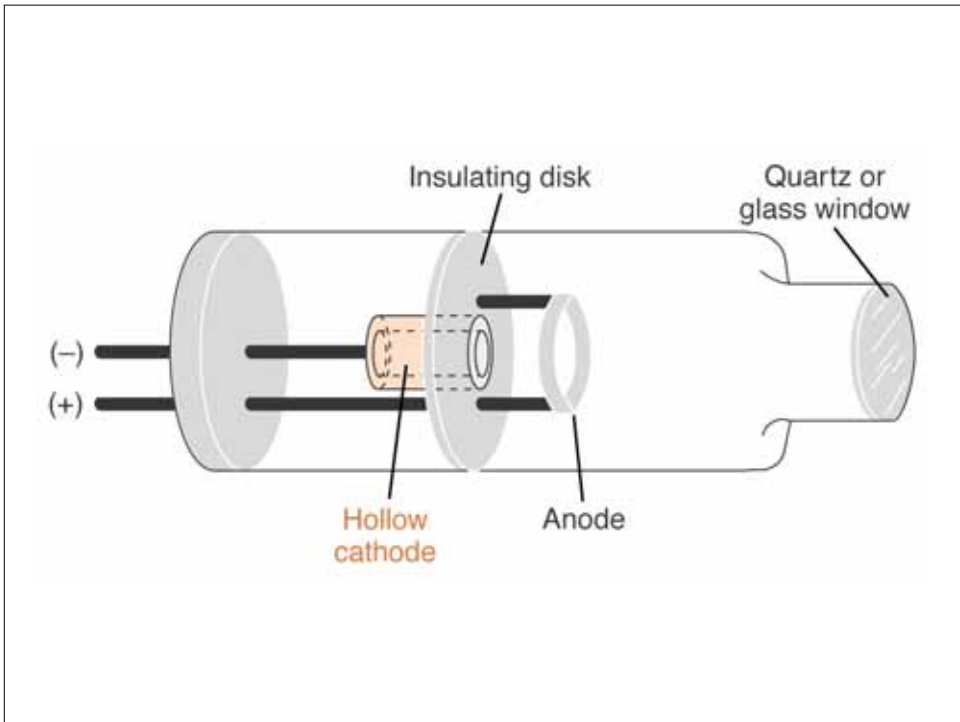
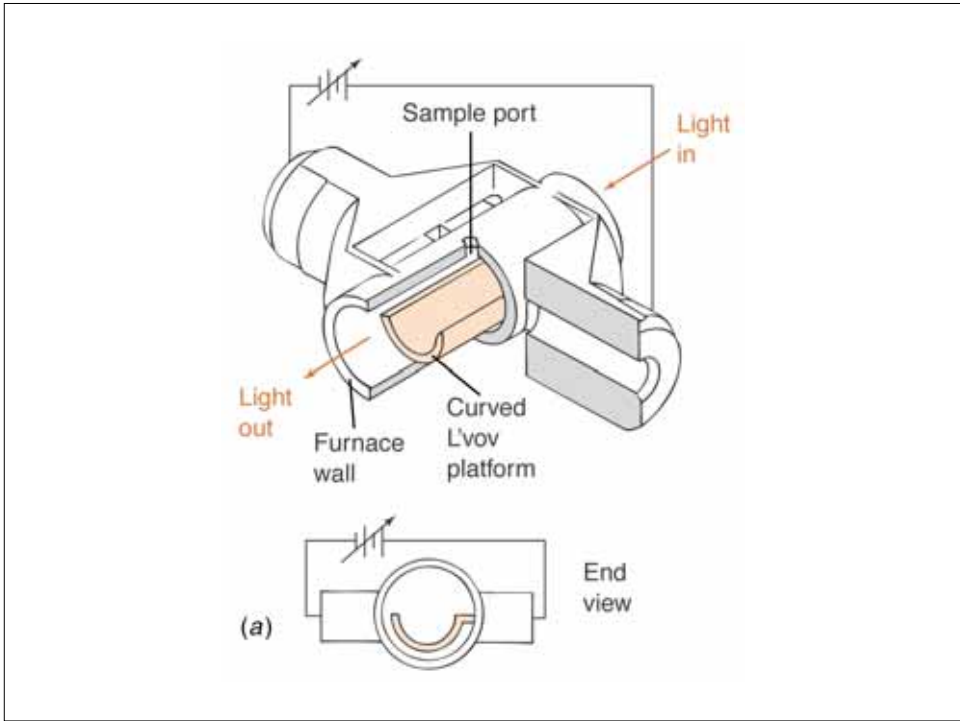


Table 21-1 Maximum flame temperatures

Fuel	Oxidant	Temperature (K)
Acetylene, HC≡CH	Air	2 400–2 700
Acetylene	Nitrous oxide, N ₂ O	2 900–3 100
Acetylene	Oxygen	3 300–3 400
Hydrogen	Air	2 300–2 400
Hydrogen	Oxygen	2 800–3 000
Cyanogen, N≡C—C≡N	Oxygen	4 800





Detection limits (ng/mL)

— Inductively coupled plasma emission
 — Flame atomic absorption
 — Graphite furnace atomic absorption
 — Inductively coupled plasma-mass spectrometry

Li 0.7 2 0.1 0.002	Be 0.07 1 0.02 0.0009	Fe 0.7 5 0.02 0.005	B 1 500 15 0.0008	C 10 — — —	N — — — —	O — — — —	F — — — —	Ne — — — —									
Na 3 0.2 0.005 0.0003	Mg 0.08 0.3 0.004 0.0003	Al 2 30 0.01 0.0002	Si 5 100 0.1 0.0001	P 7 40 000 30 0.0001	S 3 — — 0.0001	Cl 60 — — —	Ar — — — —										
K 20 3 0.1 0.0002	Ca 0.07 0.5 0.01 0.0002	Sc 0.3 40 — 0.0002	Ti 0.4 70 0.5 0.004	V 0.7 50 0.2 0.0003	Cr 2 3 0.01 0.0003	Mn 0.2 2 0.01 0.0003	Fe 0.7 5 0.02 0.0002	Co 1 4 0.02 0.0002	Ni 3 90 0.1 0.001	Cu 0.9 1 0.02 0.0005	Zn 0.6 0.5 0.001 0.003	Ga 10 60 0.5 0.006	Ge 20 200 — 0.002	As 7 200 — 0.003	Se 10 250 0.2 0.05	Br 150 — — 0.02	Kr — — — —
Rb 1 7 0.05 0.0003	Sr 0.2 2 0.1 0.0003	Y 0.6 200 — 0.0003	Zr 2 1000 — 0.0006	Nb 5 2000 — 0.0008	Mo 3 30 0.02 0.002	Tc — — — —	Ru 10 60 1 0.001	Rh 20 4 — 0.0003	Pd 4 10 0.3 0.001	Ag 0.8 2 0.005 0.0007	Cd 0.5 0.4 0.003 0.0007	In 20 40 1 0.0003	Sn 9 30 0.2 0.0009	Sb 5 40 0.1 0.001	Te 4 30 0.1 0.02	I — — — —	Xe — — — —
Ce 40 000 4 0.2 0.0003	Ba 0.6 10 — 0.0003	La 1 2000 — 0.0003	Hf 4 2000 — 0.0008	Ta 10 2000 — 0.0005	W 6 1000 — 0.002	Re 3 600 — 0.0007	Os 0.2 100 — 0.002	Ir 7 40 — 0.0004	Pt 7 100 — 0.001	Au 2 10 0.1 0.0009	Hg 7 150 2 0.0009	Tl 10 20 0.1 0.0004	Pb 10 10 40 0.0005	Bi 7 40 0.1 0.0005	Po — — — —	At — — — —	Rn — — — —
Ce 2 — 0.0003	Pr 9 8000 — 0.0002	Nd 10 1000 — 0.001	Pm — — — —	Sm 10 1000 — 0.001	Eu 0.9 20 0.5 0.0004	Gd 5 2000 — 0.001	Tb 6 500 0.1 0.0002	Dy 2 30 — 0.0008	Ho 2 40 — 0.0002	Er 0.7 30 2 0.0007	Tm 2 900 — 0.0002	Yb 0.3 4 — 0.001	Lu 0.3 300 — 0.0002				
Th 7 — 0.0003	Pa — — —	U 60 40 000 — 0.0005	Np — — — —	Pu — — — —	Am — — — —	Cm — — — —	Bk — — — —	Cf — — — —	Es — — — —	Fm — — — —	Md — — — —	No — — — —	Lr — — — —				

— Requires N₂/C₂H₂ flame and is therefore better analyzed by inductively coupled plasma
 — Best analyzed by emission

