

Sulfuric acid; H₂SO₄ is a strong mineral acid. Its uses are ore processing, fertilizer manufacturing, oil refining, wastewater processing, and chemical synthesis. If water is added to this acid it will boil and spit dangerously.

Sulfuric acid is a strong mineral acid. It's soluble in water because it's fully miscible. The boiling point is 338 °C and melting point is 10 °C. The molar mass is 98.07848 g mol⁻¹

Sulfuric acid is a strong mineral acid. Its chemical formula is H₂SO₄. It is one of the chemical industry's top products. It can be used in ore processing, fertilizer manufacturing, oil refining, and much more.

Sulfuric acid is a corrosive mineral acid used in labs, batteries, and manufacturing. The acid reacts with most metals to make sulfate and hydrogen gas and reacts violently when water is added.

Sulfuric Acid (H₂SO₄) is a strong mineral acid that is colorless when pure. This chemical is used as a chemical intermediate to manufacture other chemicals and cleaning metal surfaces.

The formula for sulfuric acid is H₂SO₄. The molar mass of sulfuric acid is 98.07848 g mol. Sulfuric acid looks colorless and clear. It is a strong mineral acid.

Sulfuric Acid, H₂SO₄, is a strong mineral acid that is soluble in water. It is a colorless oily liquid that is extremely corrosive and sometimes called oil of vitriol.

sulfuric acid - chemical formula H₂SO₄ is a mineral acid that is soluble in water. Its hydration reaction is very exothermic, and therefore the acid should be added to the water, because of the densities of the liquids.

Sulfuric acid, H₂SO₄, is an oily liquid that is colorless to dark brown. It is corrosive and inorganic. Sulfuric acid is used to manufacture many chemicals and materials including explosives and paints

Sulfuric acid a clear, colorless to brownish, dense, oily, corrosive, water-miscible liquid, H₂SO₄, usually produced from sulfur dioxide: used chiefly in the manufacture of fertilizers, chemicals, explosives, and dyestuffs and in petroleum refining.

Sulfuric acid, H_2SO_4 , is a chemical compound composed of two hydrogen atoms, one sulfur atom, and 4 oxygen atoms. Sulfuric acid can be synthesized in different concentrations, and has diverse functions depending on the concentration

Sulfuric acid, H_2SO_4 , is a colorless, highly corrosive oily. It is an oxidizing and dehydrating agent. Sulfuric acid is used in petroleum refining and in the manufacture of fertilizers, paints, and explosives.

Sulfuric acid is a sturdy mineral acid that is soluble in water. The chemical was probably discovered by Jabir ibn Hayyan, an 8th century alchemist. Another name for it, is oil of vitriol.

Sulfuric acid, H_2S_4 , is a strong mineral acid and is soluble in water at all concentrations. Some of its uses include ore processing, fertilizer manufacturing, oil refining, wastewater processing, and chemical synthesis.

Sulfuric acid, or H_2SO_4 , is an extremely powerful acid that is always soluble in water. The main uses of sulphuric acid are ore processing, fertilizer manufacturing, oil refining, wastewater processing, and chemical synthesis.

Sulfuric acid (H_2SO_4) also known as oil of vitriol has a molar mass of 98.07848 g mol⁻¹. It is a clear colorless liquid that has no smell

Sulfuric acid is a strong acid that is soluble with all concentrations of water. It is found in acid rain when sulfur dioxide is oxidated in the presence of water.

Sulfuric acid, H_2SO_4 , is a clear, dense, oily, corrosive, water-miscible liquid, that is usually produced from sulfur dioxide. Sulfuric acid is used chiefly to manufacture fertilizer, chemicals, etc.

Sulfuric acid is a strong mineral acid that is soluble in water at all concentrations. Sulfuric acid contains 2 hydrogen atoms, 1 sulfur atom, and 4 oxygen atoms.

Sulfuric Acid H_2SO_4 - A highly corrosive colorless to dark brown liquid. Used in applications such as paints, fertilizers, and detergents. It reacts with water to form hydrates and can also dehydrate in a concentrated form.

Sulfuric Acid, H_2SO_4 , is a clear liquid that is soluble in water. It was discovered by the alchemist Jabir ibn Hayyan which later led to the growth and development of manufacturing fertilizers and oil refineries.

Sulfuric acid is a strong mineral acid. Sulfuric acid has many applications and is one of the top products of the chemical industry. Principal uses include, fertilizer manufacturing, oil refining, wastewater processing, and chemical synthesis.

Sulfuric acid (H_2SO_4) is a strong mineral. Anhydrous sulfuric acid is a very polar liquid. If a sulfuric acid concentrate is mixed with water, it can be very dangerous.

Sulfuric Acid; H_2SO_4 it is a strong mineral acid that is soluble in water. Its principles are fertilizer manufacturing, oil refining, waste water processing and chemical synthesis. Many proteins are made from this acid such as cysteine and methionine

Sulfuric Acid: H_2SO_4 ; Sulfuric acid is a clear, colorless to brownish and very corrosive, It is usually produced from sulfur dioxide: It is widely used in the manufacture of fertilizers, chemicals, explosives, and dyestuffs and in petroleum refining.

Sulfuric acid: It is produced from sulfur, oxygen, and water. A strong mineral acid: colorless, oily and inorganic. It has a melting point of $10^\circ C$ and a boiling point of $338^\circ C$.

Sulfuric Acid, H_2SO_4 is a chemical compound made up of two hydrogen atom, one sulfur atom, and four oxygen atoms. Sulfuric acid is a strong acid, soluble in water, very polar and is an excellent solvent. Its applications include the production of fertilizers and detergents. It is used extensively in the steel industry, auto industry and the chemical industry.

Sulfuric acid, H_2SO_4 , is a corrosive, colorless acid that is made from burning sulfur, and oxygen. It is used to evaporate water, dissolve metals, and to add phosphorus to soil.

Sulfuric Acid: H_2SO_4 is the chemical formula for sulfuric acid, also known as oil of vitriol which was named by alchemist Jabir ibn Hayyan. Sulfuric Acid is used to dry fruit, as it is a very good dehydrating agent. Sulfuric Acid should always be added to water, never add water to Sulfuric Acid as it can boil and spew dangerously.

A strong mineral acid that is soluble in water at all concentrations. It is used in ore processing, fertilizer manufacturing, oil refining, waste water processing, and chemical synthesis.

Sulfuric acid; Sulfuric acid, H_2SO_4 , is a highly corrosive, dense, oily liquid; colorless to dark brown depending on its purity and used to manufacture a wide variety of chemicals and materials including fertilizers, paints, detergents, and explosives.

Sulfuric Acid, H_2SO_4 , is a strong mineral acid. As an acid, it reacts with most bases to produce a sulfate, and It has the ability to dissolve in water at all levels of concentrations

Sulfuric Acid- H_2SO_4 sulfuric acid is a strong mineral acid. It is soluble in water at all concentrations. Once known as oil of vitriol. Although 100% sulfuric acid can be produced only 98% is produced.

Sulfuric acid, H_2SO_4 , is a strong mineral acid. It is soluble in water at all concentrations. The chemical's probable discoverer is the 8th-century alchemist Jabir ibn Hayyan.

Sulfuric acid; Sulfuric acid is one of the top products in chemical industry. Sulfuric acid is also soluble in water at all concentrations. This acid is also found in acid rain.

Sulfuric acid; Sulfuric acid, H_2SO_4 , is a strong mineral acid. It is soluble in water at all concentrations. Principal uses include ore processing, fertilizer manufacturing, oil refining, wastewater processing, and chemical synthesis.

Sulfuric acid is a strong mineral acid. It is soluble in water at all concentrations. IT has many application such as in chemical uses. In 2001, the world used up to 165 million tonnes.

Sulfuric Acid is a strong mineral acid. It's mainly used for ore processing, fertilizer manufacturing, oil refining, wastewater processing and chemical synthesis. It's usual appearance is a clear, colorless, odorless liquid.