












Chemical Segregation and Storage Table





Adapted from Prudent Practices in the Laboratory: Handling and Disposal of Chemicals, National Research Council, 1995, University of Texas/Health Science at Houston and Boston University Environmental Health and Safety.

CLASS OF CHEMICAL	COMMON CHEMICAL EXAMPLES	ADDITIONAL CONCERNS & RECOMMENDATIONS	COMMON INCOMPATIBLE CHEMICAL TYPES	POSSIBLE REACTION IF MIXED/HEALTH CONCERNS
Corrosive Acid-Organic 	<ul style="list-style-type: none"> Acetic Acid Butyric Acid Formic Acid Glacial Acetic Acid Picric Acid Propionic Acid Trifluoroacetic Acid 	<p>Store in ventilated corrosives cabinet on protected shelving using secondary containment, keep away from water sources.</p> <p>*Do not store under the sink. *Do not store acids on metal shelving.</p>	<ul style="list-style-type: none"> Bases Cyanides Flammable Liquids Flammable Solids Inorganic Acids Oxidizers Poisons/Toxins Sulfides 	<ul style="list-style-type: none"> Gas Generation Heat Violent Reaction <p>*DO NOT POUR WATER INTO ACID</p>
Corrosive Acids-Inorganic 	<ul style="list-style-type: none"> Chromic Acid Hydrochloric Acid Hydrofluoric Acid Nitric Acid Perchloric Acid Phosphoric Acid Sulfuric Acid 	<p>Store concentrated Nitric acid (≥68%) and Sulfuric acid (≥93%) in a secondary container. Store in a corrosive cabinet labeled "Acid" or on shelving using secondary containment.</p> <p>*Do not store under the sink. *Do not store acids on metal shelving. *Hydrofluoric acid should be stored in an area accessible only by authorized personnel; do not store in glass; use plastic containers and secondary containment.</p>	<ul style="list-style-type: none"> Bases Cyanides Flammable Flammable Solids Liquids Organic Acids Oxidizers Poisons/Toxins Sulphides 	<ul style="list-style-type: none"> Gas Generation Heat Violent Reaction <p>*DO NOT POUR WATER INTO ACID</p> <p>*Perchloric acid vapor can form explosive compounds within fume hood ducts. *Hydrofluoric acid can result in severe burns to the skin and lungs</p>
Corrosive/Bases-Organic/Caustic 	<ul style="list-style-type: none"> Diamine Hydroxylamine Tetramethylethylamine Triethylamine 	<p>Store in a separate cabinet, preferable with ventilation, corrosive cabinet, or storage with a secondary container, away from potential water sources (DO NOT store under the sink).</p>	<ul style="list-style-type: none"> Acids Flammable liquids Flammable solids Inorganic Bases Poisons/Toxins 	<ul style="list-style-type: none"> Gas Generation Heat Violent Reaction
Corrosive/Bases Inorganic/Caustics 	<ul style="list-style-type: none"> Ammonium Hydroxide Calcium Hydroxide Potassium Hydroxide Sodium Hydroxide 	<p>Store in a separate cabinet, preferably with ventilation, corrosive cabinet, or storage area with a secondary container, away from potential water sources (DO NOT store under the sink).</p> <p>Store solutions of inorganic hydroxides in labeled polyethylene containers.</p>	<ul style="list-style-type: none"> Acids Flammable liquids Flammable solids Organic Bases Poisons/Toxins 	<ul style="list-style-type: none"> Gas Generation Heat Violent Reaction
Flammable Solids 	<ul style="list-style-type: none"> Carbon Charcoal Magnesium Paraformaldehyde Phosphorus 	<p>Keep in a dry, cool area away from oxidizers and corrosives</p>	<ul style="list-style-type: none"> Acids Bases Oxidizers Poisons/Toxins 	<ul style="list-style-type: none"> Fire Hazard Violent Reaction



CLASS OF CHEMICAL	COMMON CHEMICAL EXAMPLES	ADDITIONAL CONCERNS & RECOMMENDATIONS	COMMON INCOMPATIBLE CHEMICAL TYPES	POSSIBLE REACTION IF MIXED/HEALTH CONCERNS
Flammable Liquids 	<ul style="list-style-type: none"> Acetonitrile Acetone Acetone liquids with flashpoints < 100 F Benzene Diethyl Ether Ethanol Ethyl Acetate Glacial Acetic Acid Methanol Tetrahydrofuran Toluene Xylene 	<p>Flammable storage cabinet or refrigerator rated for flammable/hazardous storage/explosion-proof. *Peroxide-forming chemicals must be dated upon delivery and opening (two dates).</p>	<ul style="list-style-type: none"> Acids Bases Reactives Poisons/Toxins 	<ul style="list-style-type: none"> Heat Fire Hazard Violent Reaction
Poisons/Toxins 	<ul style="list-style-type: none"> Acrylamide Carbon Tetrachloride Chloroform Cyanides Ethidium Bromide Formamide Heavy metal compounds (e.g., Cadmium, Mercury, Osmium, Oxalic Acid, Phenol, Formic Acid) *Hydrofluoric Acid - Hydrofluoric Acid is a highly acute poison 2- Mercaptoethanol Phenol Sodium Azide 	<p>Store in a dark, dry, ventilated, cool area in an unbreakable chemically resistant secondary container (polyethylene)</p> <p>* Store volatile toxins with evaporation rate above 1.0- (ether =1.0) in flammable cabinet.</p> <p>Store non-volatile liquid poisons in a refrigerator or cabinet; amounts less than 1 liter can be stored in a cabinet above bench level, ONLY if the cabinet has sliding doors (not swinging).</p>	<ul style="list-style-type: none"> Acids Bases Corrosives Flammable liquids Oxidizers Reactives <p>Please consult Environmental Health and Safety (EH&S) for assistance prior to the use of poisonous/toxic chemicals.</p> <p>*Hydrofluoric Acid should be stored in an area accessible only by authorized personnel; do not store in glass; use plastic containers and secondary containment.</p>	<ul style="list-style-type: none"> Combustion Explosion Hazard Fire Hazard Generation of Toxic and Flammable Gas Heat Violent Reaction <p>Chloroform explosively reacts with chemically reactive metals (e.g., Aluminum or Magnesium powder, Sodium, and Lithium), Strong Oxidizers, Strong Caustics (e.g., Alkalis), and decomposes in sunlight.</p>
Explosives 	<ul style="list-style-type: none"> Ammonium Nitrate Benzoyl Peroxide Diazoisbutylnitrile Nitro Urea Picric Acid Trinitroaniline Trinitrobenzene Trinitrobenzoic Acid Trinitrophenol Trinitrotoluene Urea Nitrate 	<p>Store in a secure location away from other chemicals; store in an area away from friction or shock.</p>	<p>Please consult the Safety Data Sheets (SDSs) and EH&S prior to the use of any explosive.</p>	<ul style="list-style-type: none"> Explosion Hazard Friction Heat Shock Violent Reaction
Oxidizers 	<ul style="list-style-type: none"> Ammonium Persulfate Benzoyl Peroxide Bromates Chlorates Ethyl Acetate Ferric chloride Iodine Nitrates Peroxides Perchlorates Permanganates Potassium Dichromate Sodium Hypochlorite Super Oxides 	<p>Store in a secondary containment separately from combustibles and flammable materials.</p>	<ul style="list-style-type: none"> Combustibles Flammable Organic Materials Reducing Agents 	<ul style="list-style-type: none"> Fire hazard Gas Generation Toxic Gas








CLASS OF CHEMICAL	COMMON CHEMICAL EXAMPLES	ADDITIONAL CONCERNS & RECOMMENDATIONS	COMMON INCOMPATIBLE CHEMICAL TYPES	POSSIBLE REACTION IF MIXED/HEALTH CONCERNS
Peroxide Formers 	<ul style="list-style-type: none"> Acetals and Ketals, especially Cyclic Ethers and those with primary and/or secondary Alkyl groups Aldehydes (e.g., Acetaldehyde, Benzaldehyde) Acrylonitrile Butylated Hydroxytoluene (BHT) Dienes Tetrahydrofuran Dioxane Isopropyl Alcohol Ethers (e.g., Diethyl ether, Isopropyl Ether) Isopropyl Ether Vinyl and Vinylidene compounds 	<p>Store in airtight bottles, away from light and heat in a dark, cool dry area; avoid using containers with loose-fitting lids and ground glass stoppers; crystallization, discoloration, and formation or deposition of layers are signs of a peroxide former may have become shock sensitive; DO NOT use or move such containers: contact EH&S; all bottles of peroxide-forming chemicals MUST HAVE the received date marked on the container; when the bottle is first opened, the container must be marked with the date opened.</p>	<p>Always consult the SDS and EH&S prior to the use of any peroxide-forming chemical.</p>	<ul style="list-style-type: none"> Explosion Hazard Combustion (Exothermic Reaction) Shock Sensitive Violent Reaction <p>If an old or expired container of a peroxide-forming chemical or reactive is found, DO NOT move it. Contact EH&S ext. 3-3531 for assistance in disposing of the container.</p>
Water Reactive 	<ul style="list-style-type: none"> Alkali Metal Hydrides Lithium Metals Potassium Metals Sodium Borohydride Sodium Metals 	<p>Store in a dry, cool area away from potential spray from fire sprinklers and other water sources (DO NOT store under the sink) Label this area for water-reactive storage</p>	<ul style="list-style-type: none"> Aqueous solutions Oxidizers <p>Please consult the Safety Data Sheet (SDS) and EH&S prior to the use of water-reactive chemicals.</p>	<ul style="list-style-type: none"> Heat Violent Reaction
Flammable Compressed Gases 	<ul style="list-style-type: none"> Acetylene Arsine Butane Ethane Germane Hydrogen Propane Methane Silane 	<p>Handle flammable compressed gases in a chemical hood. Store in a well-ventilated area; store away from oxidizers, open flames, sparks, and other sources of heat ignition.</p> <p>Post NO SMOKING signs around storage room(s); flammable gases stored outdoors where ambient temperatures exceed 125 deg F (51.7 deg C) shall be protected from direct sunlight.</p> <p>Use a spark-proof wrench to attach regulators and make other connections; install a flame/flash arrestor at the regulator outlet flow valve.</p>	<ul style="list-style-type: none"> Oxidizers Toxic Compressed Gases 	<ul style="list-style-type: none"> Fire Hazard Explosion Hazard
Oxidizing Compressed Gases 	<ul style="list-style-type: none"> Chlorine Gas mixtures containing Oxygen higher than atmospheric concentrations Fluorine Oxygen Nitrogen oxides 	<p>Store oxidizers separately from flammable gas containers or combustible materials; the minimum separation requirement from these materials is 20 ft or a 5 ft noncombustible barrier with a fire-resistance rating of at least 30 minutes.</p>	<ul style="list-style-type: none"> Flammable Compressed Gases Toxic Compressed Gases 	<ul style="list-style-type: none"> Fire Hazard Explosion Hazard



CAL STATE LA

RISK MANAGEMENT / ENVIRONMENTAL, HEALTH & SAFETY

		<p>Clean equipment used for oxygen and nitrous oxide with oxygen compatible materials free from oils, greases, and other contaminants.</p> <p>Fluorine shall be handled in specially passivated containers and associated equipment</p>		
<p>Toxic Compressed Gases</p>  	<ul style="list-style-type: none"> Carbon Monoxide Hydrogen Chloride Hydrogen Sulfide Nitrogen Dioxide 	<p>Handle toxic compressed gases in a certified chemical fume hood.</p> <p>Indoor storage or use of toxic compressed gases shall be provided with a gas cabinet, exhaust enclosure, or gas room.</p> <p>Refer to the SDS information for additional guidance on the storage and compatibility requirements. Contact EH&S prior to working with toxic compressed gases.</p>	<ul style="list-style-type: none"> Flammable Compressed Gases Oxidizing Compressed Gases 	<p>Release of toxic gas Hydrogen Sulfide is a colorless, flammable, extremely hazardous gas with a "rotten egg" smell; Prolonged exposure may cause nausea, tearing of the eyes, headaches or loss of sleep, airway problems (bronchial constriction) in some asthma patients; possible fatigue, loss of appetite, headache, irritability, poor memory, dizziness, and slight conjunctivitis.</p>
<p>Strong Reducing Agents</p>	<ul style="list-style-type: none"> Acetyl Chloride Ferrous sulfide Maleic Acid Thionyl Chloride 	<p>Store in cool, dry, well-ventilated locations.</p> <p>Water-reactive.</p> <p>Segregate from all other chemicals.</p>	<p>Please consult with the specific SDS for the chemical and EH&S prior to working with a strong reducing agent.</p>	<p>Please consult with the specific SDS for the chemical and EH&S prior to working with a strong reducing agent.</p>

CLASS OF CHEMICAL	COMMON CHEMICAL EXAMPLES	ADDITIONAL CONCERNS & RECOMMENDATIONS	COMMON INCOMPATIBLE CHEMICAL TYPES	POSSIBLE REACTION IF MIXED/HEALTH CONCERNS
<p>Carcinogens</p> 	<ul style="list-style-type: none"> Benzidine Benzene Beta-Propiolactone Beta-Naphthylamine Carbon Tetrachloride Methylene Chloride 	<p>Label all containers as "Cancer Suspect Agents" or the equivalent.</p> <p>Store according to the hazardous nature of the chemical, using appropriate security when necessary</p>	<p>Please consult with the specific SDS for the chemical and EH&S prior to working with a carcinogen.</p>	<p>Please consult with the specific SDS for the chemical and EH&S prior to working with a carcinogen.</p>
<p>Teratogen</p> 	<ul style="list-style-type: none"> Aniline Benzene Lead Compounds Mercury Compounds 	<p>Label all containers as "Suspect Reproductive Hazard" or "Reproductive Effector"</p> <p>Store according to the hazardous nature of the chemical, using appropriate security when necessary</p>	<p>Aniline is incompatible with Nitric Acid and Hydrogen Peroxide.</p> <p>Please consult the specific SDS and EH&S prior to working with a teratogen.</p>	<p>Please consult with the specific SDS for the chemical and EH&S prior to working with a teratogen.</p>
<p>General Stock Chemicals</p>	<ul style="list-style-type: none"> Agar Most non-reactive salts Salt Butter Sodium Bicarbonate Sodium Chloride 	<p>Store on shelves, or laboratory benches or shelving preferably behind glass doors and below eye level with like chemicals</p>	<p>Please consult the specific SDS for the chemical and EH&S prior to working with general stock chemicals.</p>	<p>Please consult the specific SDS for the chemical and EH&S prior to working with general stock chemicals.</p>